





### Scheme II

In a molecule of this type one would expect the molecular ion to lose  $C_3H_4O$  and also fracture at bonds between adjacent methylene groups (2). This is in fact what happens. The major peaks below  $m/e = 135$  are at 41, 55, 69. This 69 peak corresponds to the  $\omega$ -terminal residue coming as a charged fragment.

In the nmr spectrum the resonance in the region at  $\delta$  1.6 is due to dimethyl allyl group. The peaks of  $\delta$  7.2 and 7.4 respectively must be due to 5 protons on the aromatic ring. The absorption band at  $\delta$  5.0 is a sort of a multiplet showing the presence of grouping  $HC-O-CO-A_y$ . The infra-red spectrum shows the presence of unsaturated linkage at  $1650cm^{-1}$  overlapping with aryl carbonyl group. Two other bands at about  $980$  and  $920cm^{-1}$  arise from out of plane bending motions on vinyl group. The aromatic and the aliphatic C-H stretching frequencies respectively are apparent on the higher frequency and lower frequency side of  $3000cm^{-1}$ . Bands at  $1590$  and  $1450cm^{-1}$  are associated with the presence of phenyl group. The methylene bending occurs  $1420cm^{-1}$ . Shown with good resolution the absorption bands at  $2960$  and  $2920cm^{-1}$  are assignable to C-H stretchings in methyl and methylene groups respectively. The absorption band at  $1375cm^{-1}$  is from a symmetrical deformation of the terminal methyl groups. The bands at  $1270$  and  $1230cm^{-1}$  are due to C-O stretching vibrations. The absorption bands at  $790$ ,  $770$ ,  $690$  and  $655cm^{-1}$  show the presence of five adjacent protons in the phenyl group.

Quantitative determination of isopropylidene groups (3) in the compound indicated that the molecule contained only one isopropylidene group in agreement with structure I.

Structure I required one terminal methyl in each isoprene unit. A determination showed 0.72-0.73 terminal methyl group for each isoprene unit (4). Since the terminal methyl analysis usually show 0.8 or less for each group, our results are consistent with structure I.

#### Antibacterial Screening:

The compound isolated from the petroleum ether extract was subjected for investigation of antibacterial activity. The sample and the solvent (Absolute ethanol) system sterilized previously and 1% W/v were placed especially in bores made in nutrient agar seeded with test organisms. The petri-dishes were examined for bacterial growth after twenty-four hours incubation at  $37^\circ C$  and then compared. The highest zones of inhibition were produced by the sample.

#### Microbiological Screening:

Organism	Gram type	Strain	Zone of inhibition for the compound (mm)	Zone of inhibition for the control (mm)
<i>Pseudomonas aeruginosa</i>		isolated locally	15±0.2	1.5
<i>Staphylococcus Aureus</i>	+	ATCH 10081	14.4±0.1	1.5
<i>Bacillus subtilis</i>	+	isolated locally	14.2±0.2	1.5
<i>Streptococcus Pneumoniae</i>	+	isolated locally	14.0±0.1	1.5
<i>Candida Albicans</i>	+	isolated locally	14.1±0.1	1.5
<i>Staphylococcus Aureus</i>			15±0.2 14.4	1.5

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